## PATENT COOPERATION TREATY

## **PCT**

REC'D 1 2 OCT 2005

INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY WIPO

(Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

0000053704		FOR FURTHER ACTION	See Form PCT/IPEA/416	
International application No. PCT/EP2004/007256		International filing date (day/month/year) 03.07.2004	Priority date (day/month/year) 23.07.2003	
International Patent Class A62D1/00	ssification (IPC) or nati	onal classification and IPC	23.07.2003	
Applicant				
BASF AKTIENGES	ELLSCHAFT			
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# INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

International application No. PCT/EP2004/007256

_	Box	No. I	Basis of the report
1.	With filed,	regard unles	to the <b>language</b> , this report is based on the international application in the language in which it was so otherwise indicated under this item.
	]	□ inte	port is based on translations from the original language into the following language, is the language of a translation furnished for the purposes of:  rnational search (under Rules 12.3 and 23.1(b))  lication of the international application (under Rule 12.4)  rnational preliminary examination (under Rules 55.2 and/or 55.3)
2.	With have	regard been	I to the <b>elements*</b> of the international application, this report is based on <i>(replacement sheets which furnished to the receiving Office in response to an invitation under Article 14 are referred to in this priginally filed" and are not annexed to this report):</i>
	Desc	ription	, Pages
	1-20		as originally filed
	Claim	ns, Nun	nbers
	1-15		received on 12.09.2005 with letter of 12.09.2005
	□ a	a sequ	ence listing and/or any related table(s) - see Supplemental Box Relating to Sequence Listing
3.	[] [] []	the the the the	nendments have resulted in the cancellation of: description, pages claims, Nos. drawings, sheets/figs sequence listing (specify): table(s) related to sequence listing (specify):
4.	Supp	lement the call the call the call the call any	port has been established as if (some of) the amendments annexed to this report and listed below an made, since they have been considered to go beyond the disclosure as filed, as indicated in the description, pages claims, Nos.  drawings, sheets/figs sequence listing (specify): table(s) related to sequence listing (specify):
	* I	f ite	em 4 applies, some or all of these sheets may be marked "superseded."

# INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

International application No. PCT/EP2004/007256

_	Bo	x No. II	Priority									
_		X 140. II	Triority			-						
1.	This report has been established as if no priority had been claimed due to the failure to furnish within the prescribed time limit the requested:  Copy of the earlier application whose priority has been claimed (Rule 66.7(a)).											
$\Box$ translation of the earlier application whose priority has been claimed (Rule 66.7(b)).												
2.	This report has been established as if no priority had been claimed due to the fact that the priority claim has been found invalid (Rule 64.1). Thus for the purposes of this report, the international filing date indicated above is considered to be the relevant date.											
3.	Add	litional of	bservations, if necess	arv:								
	Box	( No. V licabilit	Reasoned stateme y; citations and expl	ent und anatio	er Article	35(2) with	regard to novelty, inventive step or indust	rial				
1.	Stat	tement										
	Nov	elty (N)		Yes: No:	Claims Claims	1-15						
	Inve	entive ste	ep (IS)	Yes: No:	Claims Claims	1-15	•					
	Indu	ıstrial ap	plicability (IA)	Yes: No:	Claims Claims	1-15	·					

2. Citations and explanations (Rule 70.7):

#### Re Item V.

1 The following documents are referred to in this communication:

D1: US 3 354 084 A (KATZER MELVIN F) 21 November 1967 (1967-11-21)

D2: DATABASE WPI Section Ch, Week 198145
Derwent Publications Ltd., London, GB; Class A97, AN 1981-84248
XP002297922 &; JP 56 125066 A (OTSUKA KAGAKU YAKUHIN) 1 October 1981 (1981-10-01)

D3: US 2002/189492 A1 (VANDERSALL HOWARD L ET AL) 19 December 2002 (2002-12-19)

D4: EP 0 911 067 A (CHEMONICS FIRE TROL INC) 28 April 1999 (1999-04-28)

D5: US-A-5 849 210 D6: US-A-5 190 110

### 1. Novelty, Art. 33(2) PCT.

D1 (claim 5) refers to a composition comprising water, 0,05-2 wt.% of a water swellable acrylic polymer, 0,1-2 pts.wt. of a non-ionic solid which is titanium oxide (claim 2) and 0,01-0,3 pts.wt of a dye. The water swellable acrylic copolymers can be used "to thicken water used in fighting fires" (col. 1. I. 14-19).

Document D2 describes an aqueous solution of 2-3 wt% polyacrylic acid or polyacrylamide, water, ammonium phosphate and 0,01-0,03 wt% colorant.

The subject-matter of claims 1-15 is distinguished from the closest prior art D1-D2 in that it refers to specific opacifying agents.

### 2. Inventive step, Art. 33(3) PCT

There was no suggestion in the prior art to select the specific opacifying agents. These compositions are not derivable from any of D1-D6 whether read individually or in any combination.

Hence, an inventive step can be recognized for these claims.

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#### WHAT IS CLAIMED IS:

- A method of controlling a fire consuming a combustible object comprising a step of applying a sufficient amount of a fire-fighting composition to the combustible object to retard, suppress, or extinguish the fire, said fire-fighting composition comprising:
  - (a) 0.01% to 20%, by weight, of a superabsorbent polymer;
  - (b) 0.005% to 10%, by weight, of a colorant;
- (c) 0.005% to 10%, by weight, of an additional opacifying agent selected from the group consisting of an inorganic compound having a water solubility, in its anhydrous form of 0.0005 to 0.005 grams per 100 ml of water at 25°C, a synthetic organic homopolymer or copolymer, and mixtures thereof; and
  - (d) water.

 A method of preventing a fire from consuming a combustible object comprising a step of applying a sufficient amount of a fire-fighting composition to the combustible object to inhibit initiation of the fire, said fire fighting composition comprising:

- 20 (a) 0.01% to 20%, by weight, of a superabsorbent polymer;
  - (b) 0.005% to 10%, by weight, of a colorant;
  - (c) 0.005% to 10%, by weight, of an additional opacifying agent selected from the group consisting of an inorganic compound having a water solubility, in its anhydrous form of 0.0005 to 0.005 grams per 100 ml of water at 25°C,, a synthetic organic homopolymer or copolymer, and mixtures thereof; and
  - (d) water.
  - The method of claim 1 or 2 wherein the colorant and the opacifying agent are
    present in a weight ratio of 1 part colorant to about 0.25 to about 5 parts opacifying agent.
    - 4. The method of one of the claims 1 to 3 wherein the opacifying agent and superabsorbent polymer are present in a weight ratio of 1 part opacifying agent to 1.5 to 10 parts superabsorbent polymer.
    - 5. The method of one of the claims 1 to 4 wherein the fire is a forest fire, a grass fire, a brush fire, or a wildfire.
- 6. The method of one of the claims 1 to 5 wherein the combustible objects comprises a man-made structure.

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- 7. The method of one of the claims 1 to 6 wherein the superabsorbent polymer is present in an amount of 0.05% to 10%, by weight of the composition.
- The method of one of the claims 1 to 7 wherein the superabsorbent polymer
   comprises a polymerized α,β-unsaturated carboxylic acid, or a salt or an anhydride thereof.
- The method of one of the claims 1 to 8 wherein the superabsorbent polymer is selected from the group consisting of poly(acrylic acid), a hydrolyzed starch-acrylonitrile graft copolymer, a starch-acrylic acid graft copolymer, a saponified vinyl acetate-acrylic ester copolymer, a hydrolyzed acrylonitrile copolymer, a hydrolyzed acrylamide copolymer, an ethylene-maleic anhydride copolymer, an isobutylene-maleic anhydride copolymer, a poly(vinylsulfonic acid), a poly(vinylphosphonic acid), a poly(vinylphosphoric acid), a poly(vinylsulfuric acid), a sulfonated polystyrene, and salts and mixtures thereof.
  - 10. The method of the claims 1 to 9 wherein the superabsorbent polymer is selected from the group consisting of a poly(vinylamine), a poly(dialkylaminoalkyl (meth)acrylamide), a polyethylenimine, a poly(allylamine), a poly(allylamidine), a poly(dimethyldiallylammonium hydroxide), a quaternized polystyrene derivative, a guanidine-modified polystyrene, a quaternized poly((meth) acrylamide) or ester analog, a poly(vinylguanidine), salts thereof, and mixtures thereof.
- 11. The method of one of the claims 1 to 10 wherein the superabsorbent polymer comprises polyacrylic acid neutralized 50% to 100%.
  - 12. The method of one of the claims 9 to 11 wherein the polyacrylic acid is neutralized with sodium hydroxide, sodium carbonate, potassium hydroxide, potassium carbonate, or a mixture thereof.
  - 13. The method of one of the claims 1 to 12 wherein the colorant is present in an amount of 0.01% to 5%, by weight of the composition.
- 14. The method of one of the claims 1 to 13 wherein the colorant is a dye, a pigment, or a mixture thereof.
  - 15. The method of one of the claims 1 to 14 wherein the colorant imparts a yellow, red, orange, violet, or blue color to the composition.
- 40 16. The method of one of the claims 1 to 15 wherein the colorant is selected from the group consisting of a direct dye, a basic dye, an acid dye, a reactive dye, a solvent dye, a disperse dye, a leather dye, a natural dye, a sulfur dye, a vat dye, a

synthetic pigment, a naturally occurring pigment, a security dye, and mixtures thereof.

17. The method of claim 16 wherein the colorant comprises a direct dye.

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18. The method of claim 17 wherein the direct dye is selected from the group consisting of Direct Red 81, Direct Red 239, Direct Red 254, Direct Yellow 11, Direct Yellow 6, Direct Yellow 127, Direct Orange 102, Direct Orange 102:1, Direct Orange 116, Direct Yellow 5, and mixtures thereof.

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- 19. The method of the claims 17 or 18 wherein the colorant comprises Red 51L, Orange 80LN, or Yellow 76LN.
- The method of one of the claims 1 to 19 wherein the opacifying agent is present in an amount of 0.01% to 5%, by weight of the compositions.
  - 21. The method of one of the claims 1 to 20 wherein the opacifying agent is selected from the group consisting of calcium carbonate, a styrene-butadiene copolymer, a styrene-vinylpyrrolidone copolymer, a styrene-butadiene-acrylonitrile copolymer, an acrylic polymer, a polyvinyl acetate, a polyvinyl acrylate, a starch, a polyethylenimine, a polystyrene, a polyethylene, a polyvinyl alcohol, and mixtures thereof.
- 22. The method of one of the claims 1 to 21 wherein the opacifying agent comprises calcium carbonate.
  - 23. The method of one of the claims 1 to 22 wherein the opacifying agent comprises an emulsion or a latex of the synthetic homopolymer or copolymer.
- 30 24. The method of claim 23 wherein the opacifying agent comprises a polyethylenimine.
  - 25. The method of one of the claims 1 to 24 wherein the composition further comprises up to 10%, by weight, of a water soluble organic solvent.

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- 26. The method of one of the claims 1 to 25 wherein the composition further comprises one or more optional ingredient selected from the group consisting of a viscosity modifier, a dispersant, a pH modifier, a surfactant, and mixtures thereof.
- 40 27. The method of one of the claims 1 to 26 wherein the composition imparts a color to the combustible object.

- 28. The method of claim 27 wherein the color imparted to the combustible object substantially fades within 30 days after application of the composition.
- The method of claim 27 or 28 wherein the color imparted to the combustible object is of sufficient intensity such that a combustible object having the composition applied thereto is differentiated from the combustible object that lacks on application of the composition by a naked eye.
- 30. The method of one of the claims 1 to 29 wherein the composition is applied by ground equipment or by aerial equipment.
  - 31. A composition comprising
    - (a) 0.1% to 5%, by weight, of a superabsorbent polymer;
- 15 (b) 0.015% to 2%, by weight, of a colorant;
  - (c) 0.015% to 2%, by weight, of an additional opacifying agent selected from the group consisting of an inorganic compound having a water solubility, in its anhydrous form of 0.0005 to 0.005 grams per 100 ml of water at 25°C, a synthetic organic homopolymer or copolymer, and mixtures thereof; and
- 20 (d) water.
  - 32. The composition of claim 31 further comprising up to about 10%, by weight, of a water-soluble organic solvent.
- 25 33. The composition of claim 31 or claim 32 further comprising one or more optional ingredient selected from the group consisting of a viscosity modifier, a dispersant, a pH modifier, a surfactant, and mixtures thereof.
- 34. The composition of one of the claims 31 to 33 wherein the colorant and the opacifying agent are present in a weight ratio of 1 part colorant to 0.5 to 3 parts opacifying agent.
- The composition of one of the claims 31 to 34 wherein the opacifying agent and the superabsorbent polymer are present in a weight ratio of 1 part opacifying agent to 2 to 4 parts superabsorbent polymer.